

# THE UNIVERSITY OF THE WEST INDIES

ST. AUGUSTINE, TRINIDAD & TOBAGO, WEST INDIES

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## Test Report

### Determination of Alcohol Content (%) in Sanitizer Products

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## INTRODUCTION

One (1) sanitizer sample was delivered to the Department of Chemistry of The University of the West Indies, St Augustine by the Client for determination of percentage alcohol content. This report presents the result for the following sample:

Product Name	Date Delivered
Hansan W 62%	29.06.2021

## METHOD:

### *Chemicals/Reagents*

Isopropanol /IPA (BDH, 99.0%) and *n*-butanol (Sigma Aldrich, 99.5%) were used for standard and sample preparations. The standards and samples were diluted in deionized water to specified concentrations.

### *Instrumentation*

A Shimadzu GC-2014 Gas Chromatograph equipped with split/splitless injector (SPL) and flame ionization detector (FID) was used for this analysis and the data were acquired, analyzed and reported using the *GCsolution* software. The method parameters are shown in Table 1 below.

**Table 1.** Methods parameters for determination of alcohol content in sanitizers

GC system	Shimadzu GC-2014 with SPL and FID
Column	SHRX-5, 15m x 0.25mm x 0.25µm
Injector Mode	Split at 1:90 ratio
Injection Volume	2.0 µL
Carrier Gas	Helium (He)
Flow mode	Constant linear velocity of 12cm/sec
Column Temperature	35°C — 50°C @ 1°C/min
Injection Port Temperature	275°C
FID Temperature and Carrier Gas Flow Rate	300°C, Helium 39.4mL/min

### *Preparation of Calibration Curves*

Quantification was obtained using the Internal Standard Method. Stock concentrations (20%, 40%, 60% and 80%) of IPA solutions were prepared in deionized water. Working standard solutions of each concentration (20%, 40%, 60% and 80%) were prepared in separate flasks by adding 5 mL IS (5% *n*-butanol), 5 mL of the respective IPA stock solution and diluting to 50mL with deionized water. 2 µL of each working solution was injected into the instrument and the peak areas of IPA and IS were obtained. The IPA to *n*-butanol peak area ratio was calculated for each working standard and a plot of peak area ratio against % concentration was generated. The calibration curve obtained is shown in Figure 1.

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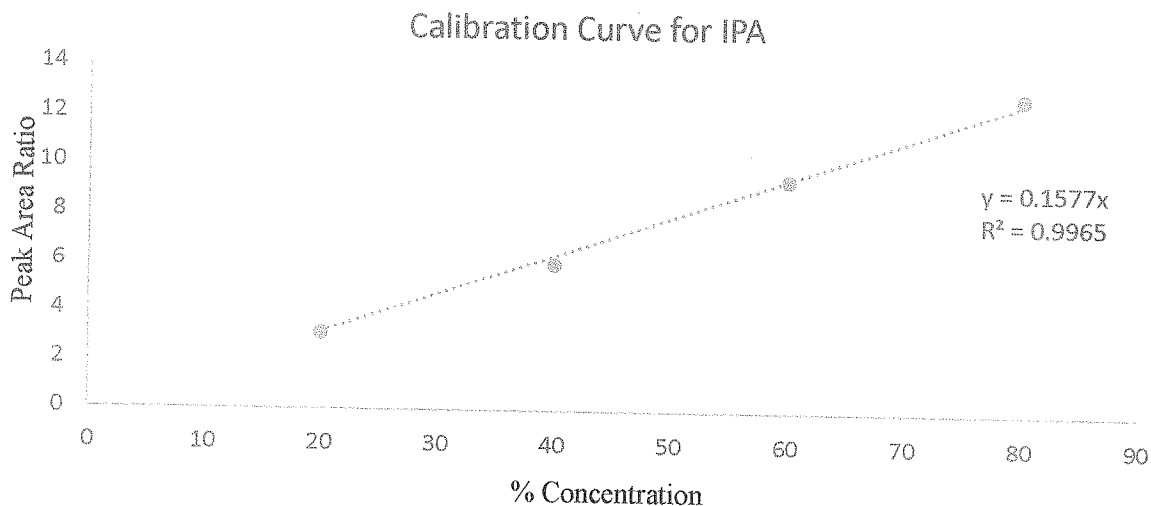


Figure 1. Calibration Curve for Isopropanol

*Preparation of Sanitizer Samples*

5 mL of sample and 5 mL of IS (5% *n*-butanol) was diluted to 50mL with deionized water. 2  $\mu$ L of the diluted sample was injected into the instrument and the peak areas of IPA and IS were obtained. The IPA to *n*-butanol peak area ratio was calculated and the % IPA concentration was determined using the slope of the graph (0.1577) according to the following equation:

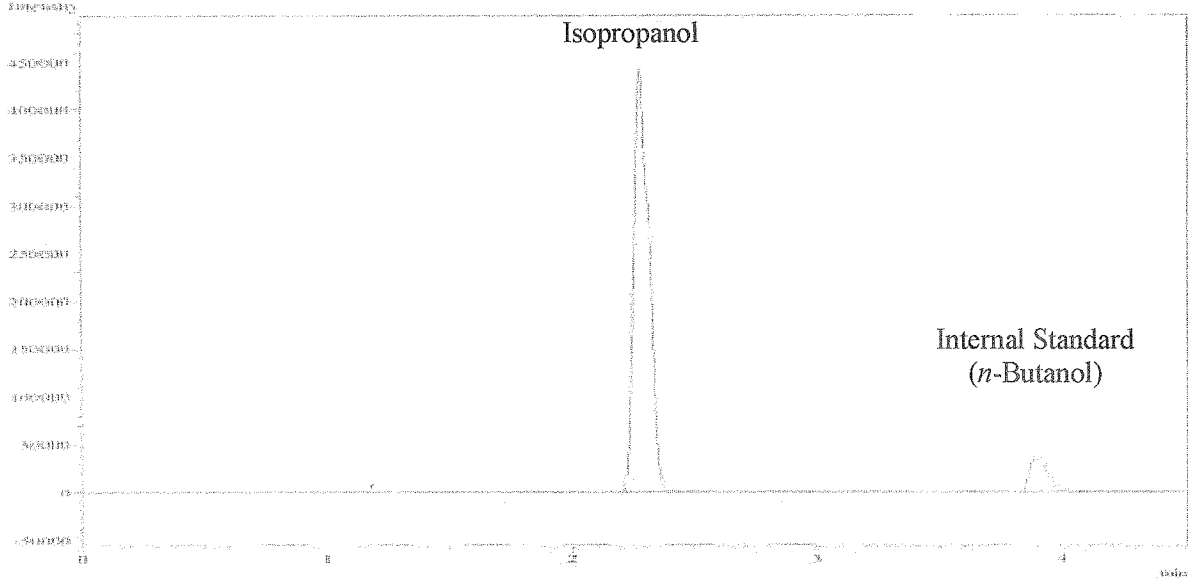
$$\% \text{ Concentration} = \frac{\text{Peak Area Ratio for Sample}}{\text{Slope of Graph}}$$

Analyses were conducted in duplicate and results reported as mean  $\pm$  standard deviation.

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**RESULTS<sup>1</sup>:**

A sample chromatogram is shown in Figure 2 below:



**Figure 2. Sample Chromatogram**

The percentage alcohol (Isopropanol) content (v/v) in the sanitizer sample is reported in Table 2 below:

**Table 2. Percentage alcohol content in sanitizer sample**

Sample I.D.	Isopropanol concentration (% v/v) (Mean ± Standard Deviation)
Hansan W 62%	62.20 ± 3.00

*End of test report*

<sup>1</sup> These test results relate to the specific sample(s) analysed and are intended to be used for information purposes only.

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